

# Abschn. 5.5 Numerische Durchführung

DFT:  $E_g = E[n]$ ,  $n(\vec{r})$  Elektronendichte

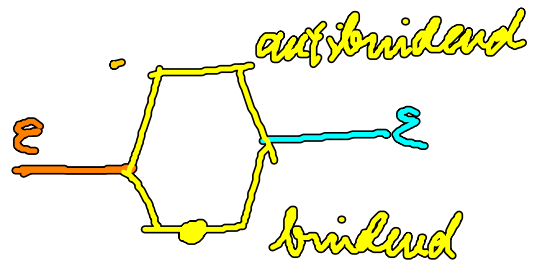
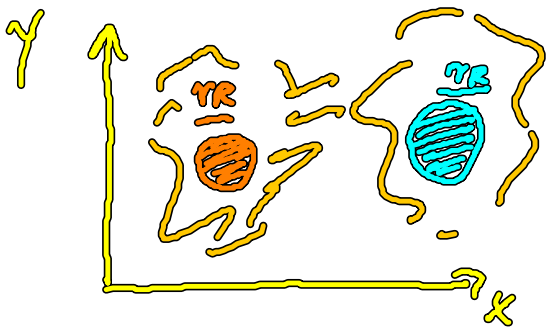
aus  $N$ -Elektronenpotential  $v(\vec{r}) = v(\vec{r}, \vec{R}_1, \vec{R}_2, \dots, \vec{R}_N)$

$E_g(\vec{R}_1, \vec{R}_2, \dots, \vec{R}_N) \Rightarrow$

1) Molekülstruktur Min.  $E_g$ :  $\frac{\partial E_g}{\partial \vec{R}_1} = 0$ ;  $\frac{\partial E_g}{\partial \vec{R}_2} = 0$ , ...  $\frac{\partial E_g}{\partial \vec{R}_N} = 0$   
Verfahren des konjugierten Gradienten

2)  $\frac{\partial E_g}{\partial \vec{R}_j} = -\vec{F}_j$  aus  $M_j \ddot{\vec{R}}_j(t) = \vec{F}_j$ ;  $\ddot{\vec{R}}_j(t) = \frac{1}{M_j} \vec{F}_j = -\frac{1}{M_j} \frac{\partial E_g}{\partial \vec{R}_j}$

$\Delta t$  mit  $\ddot{\vec{R}}_j(t) \approx \frac{1}{\Delta t^2} (\vec{R}_j(t-\Delta t) - 2\vec{R}_j(t) + \vec{R}_j(t+\Delta t))$



## Kapitel 6 Anwendungen der DFT